

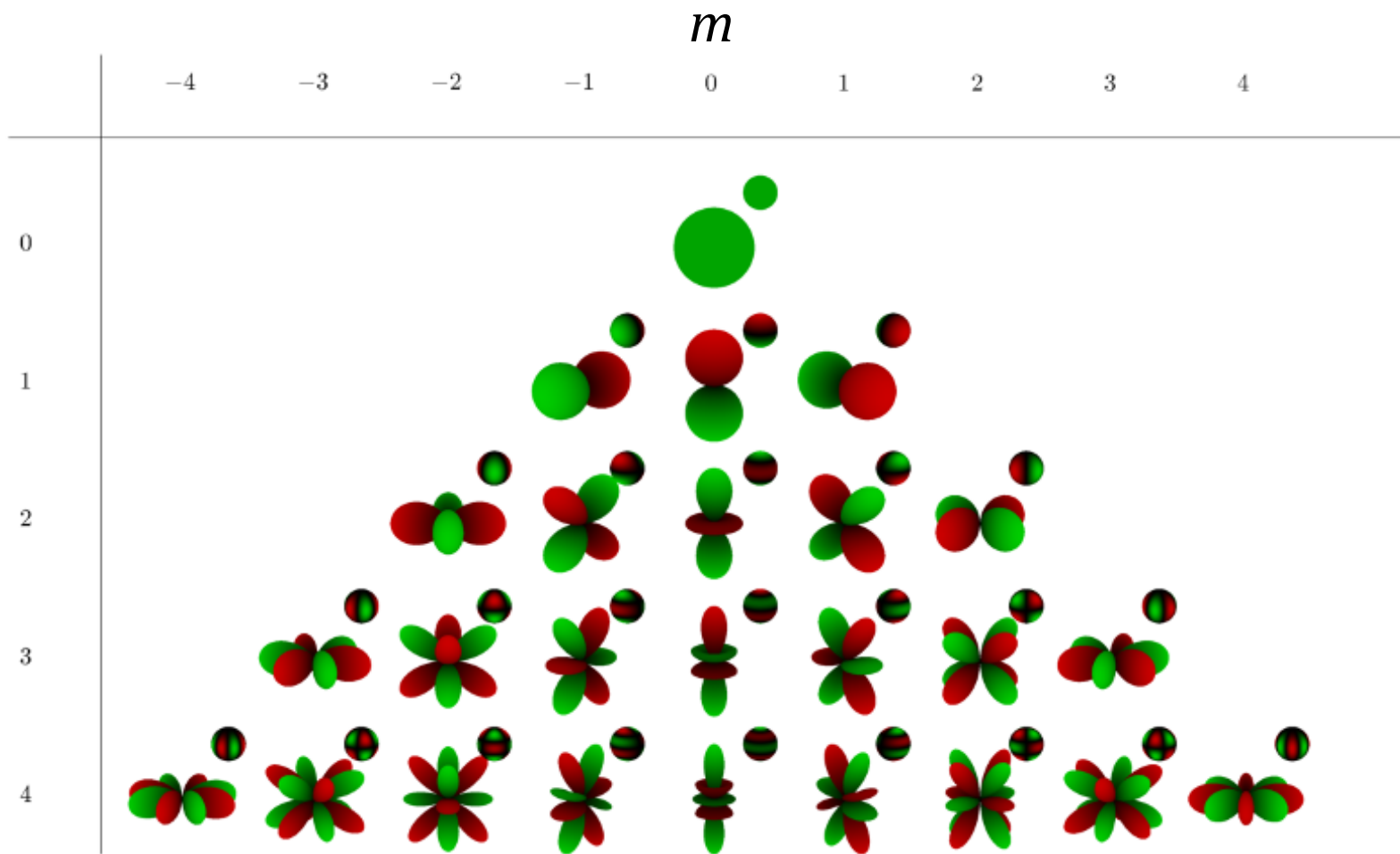
Take another look at the spatial distributions of the wavefunction solutions to the rigid rotor Hamiltonian - the spherical harmonics.

A. What information does the quantum number J encode?

B. What information does the quantum number m encode?

J

$$Y_J^m(\theta, \phi)$$



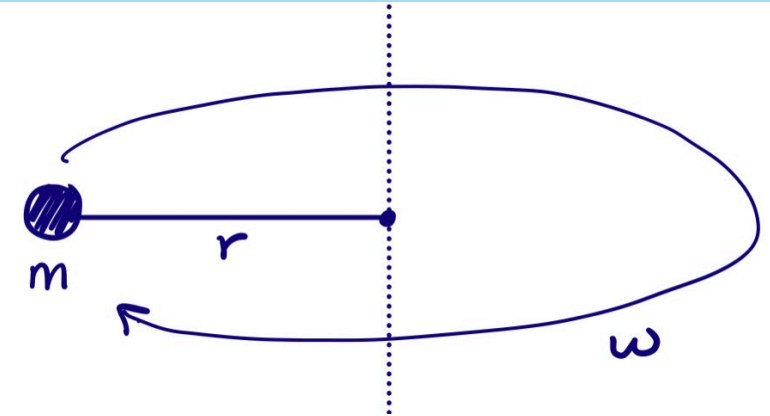
CHM 305 The Quantum World

Lecture 11: Molecular Spectroscopy

McQuarrie Ch. 5, 6

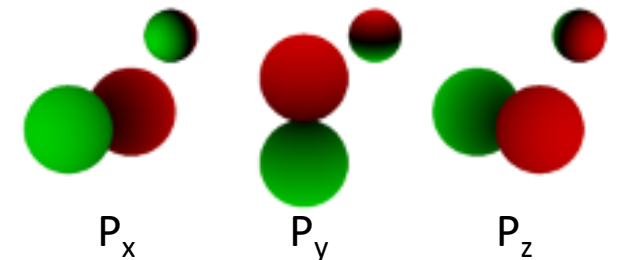
Last time we talked about: angular momentum

- Lay out definitions for classical circular motion and angular momentum
- Discuss rotations of classical and quantum mechanical rigid bodies
- Write down the Schrodinger equation and its solutions for the quantum rigid rotor in spherical coordinates
- Make connections to rotations of diatomic molecules and the hydrogen atom



$$\hat{H} = -\frac{\hbar^2}{2\mu} \nabla^2$$

$$\hat{H}\psi(\theta, \phi) = E\psi(\theta, \phi)$$



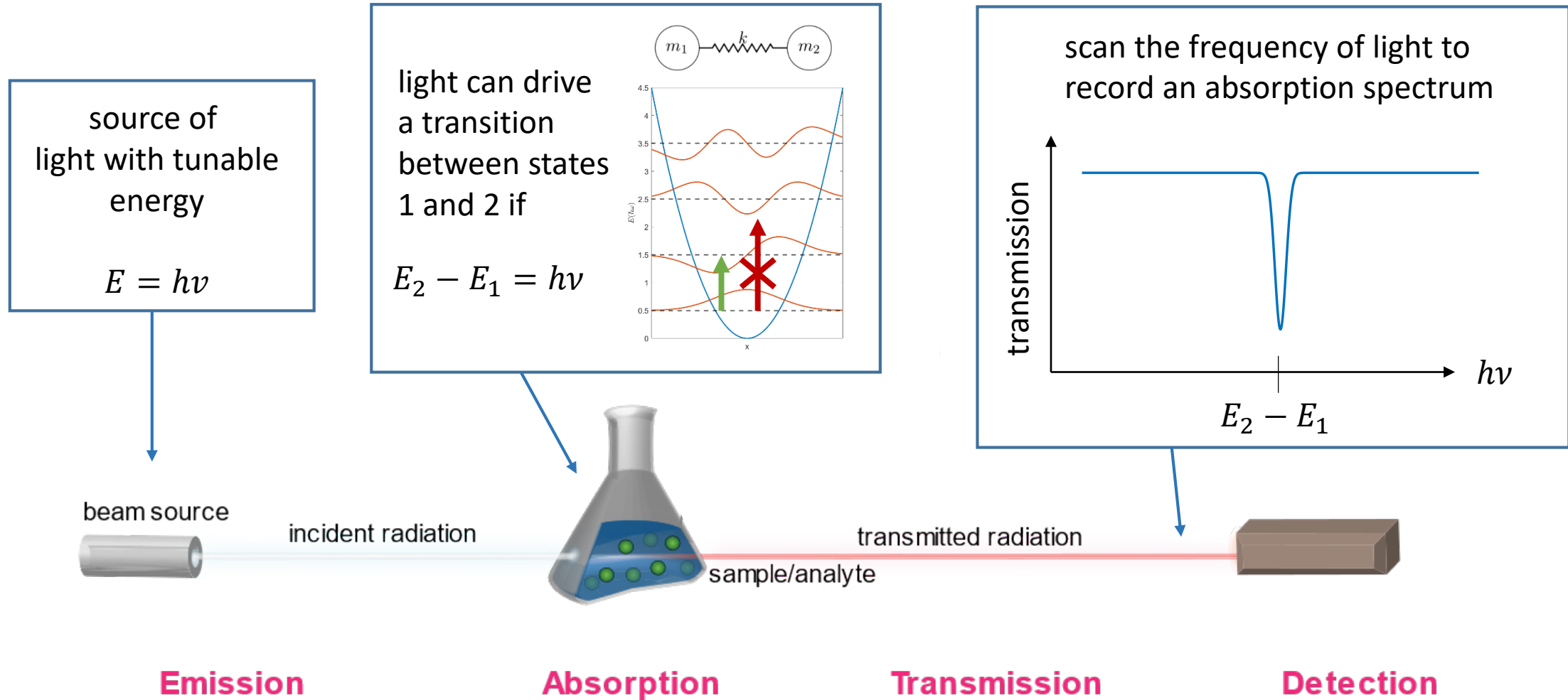
Road map for today's lecture

Overall goal: use harmonic oscillator and rigid rotor to model the frequencies of light that molecules absorb

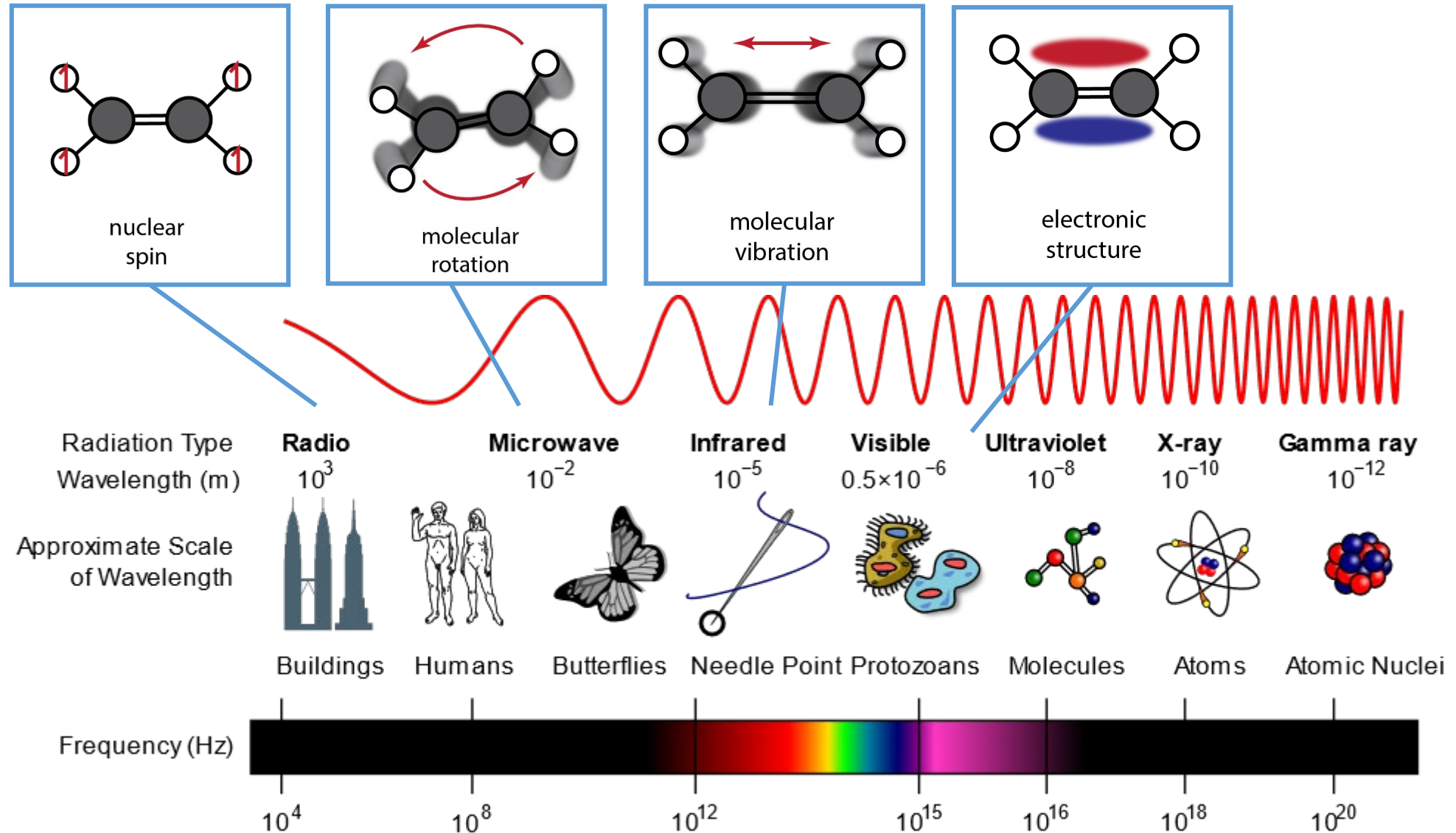
1. Harmonic oscillator \leftrightarrow vibrational/infrared spectroscopy
transition energies, selection rules
beyond the HO approximation (anharmonicity)
2. Rigid rotor \leftrightarrow rotational/microwave spectroscopy
transition energies, selection rules
beyond the rigid rotor approximation (centrifugal distortion)
2. Harmonic oscillator-rigid rotor \leftrightarrow rovibrational spectroscopy
and rotation-vibration interaction



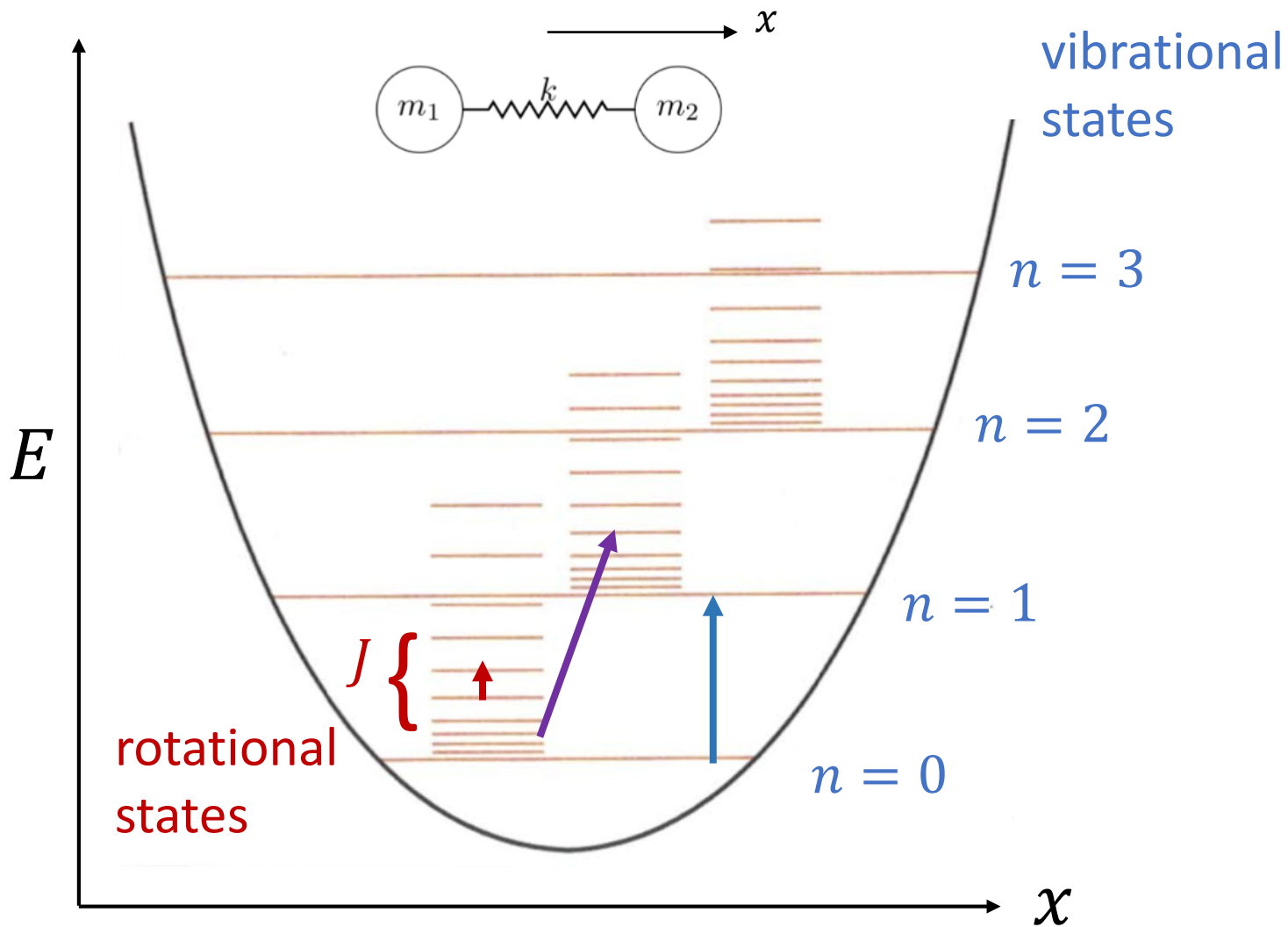
The basic principle of absorption spectroscopy



Different frequencies of light drive different molecular transitions



Rovibrational transitions



Vibrations

- Harmonic oscillator model
- Transitions driven by infrared light = IR spectroscopy

Rotations

- Rigid rotor model
- Transitions driven by microwave light = microwave spectroscopy

Can also drive “rovibrational” transitions that change both n and J quantum numbers